

# Esha Summer Project

## 2023

By: Aaratrika S.

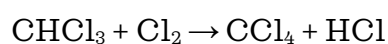
Grade- 12

Chirec International School

Chemistry:

Carbon and its compounds

- 1) The following chemical reaction shows the addition of chlorine gas to hydrocarbon in the presence of sunlight.



How does chlorine react to a hydrocarbon compound in the presence of sunlight?

- (a) It adds hydrogen to the compound
  - (b) It adds an oxygen atom to the compound
  - (c) It substitutes hydrogen atom from the compound - correct answer
  - (d) It breaks double and triple bonds into a single bond
- 2) Which of the following is the molecular formula of cyclobutane?
- (a)  $\text{C}_4\text{H}_8$  - correct answer
  - (b)  $\text{C}_4\text{H}_6$
  - (c)  $\text{C}_4\text{H}_{10}$
  - (d)  $\text{C}_4\text{H}_4$

3) Which of the following is the molecular formula of pentane?

(a)  $C_6H_{14}$

(b)  $C_5H_{12}$  - correct answer

(c)  $CH_4$

(d)  $C_3H_8$

4) Which is denatured spirit?

- (a) Ethanol only
- (b) Ethanol and methanol - correct answer
- (c) Propanol only
- (d) Methanol only

5) Soaps are formed by saponification of

- (a) Alcohols
- (b) Glycosides
- (c) Simple esters - correct answer
- (d) Carboxylic acids

6) What helps the dirt to rise away?

- (a) Suspension of the dirt in the micelles - correct answer
- (b) A collection of water molecules in the centre of the micelle
- (c) The attraction between the ionic end and the dirt to remove it.
- (d) Mixing of the soap molecules along with the dirt to make it heavier.

7) How many single bonds are present in Ethane?

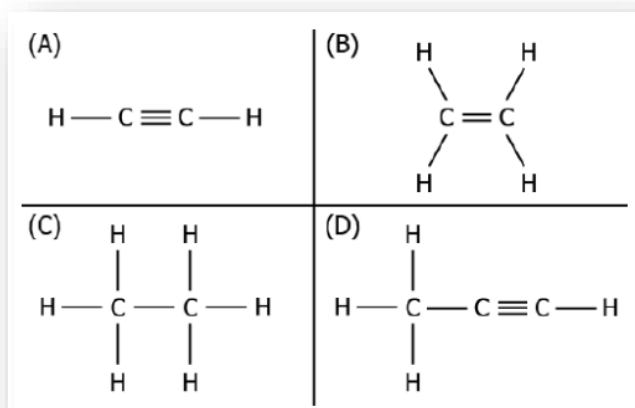
(a) 7, all carbon-carbon bonds

(b) 6, all carbon-carbon bonds

(c) 1 C-H bond and 6 C-C bonds

(d) 1 C-C bond and 6 C-H bonds - correct answer

8) Which of these compounds can be classified as alkynes?



(a) Only (A)

(b) Only (B)

(c) Both (A) and (D) - correct answer

(d) Both (B) and (C)

9) Which functional group is present in the compound?

(a) Alcohol

(b) Aldehyde - correct answer

(c) Carboxylic acid

(d) Ketone

10) A hydrocarbon should have a minimum of \_\_\_\_\_ carbon atoms to show isomerism.

(a) Three

(b) Four

(c) Five

(d) Six - correct answer

ANSWER KEY:

- |     |     |
|-----|-----|
| 1)  | (c) |
| 2)  | (a) |
| 3)  | (b) |
| 4)  | (b) |
| 5)  | (c) |
| 6)  | (a) |
| 7)  | (d) |
| 8)  | (c) |
| 9)  | (d) |
| 10) | (b) |